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### Synthesis and Physical Properties of Ti-O Monolayer Superconductors: $T1Ba_{1+x}La_{1-x}CuO_5$ and $TiBa_2Y_{1-x}Ca_xCu_2O_7$

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## SYNTHESIS AND PHYSICAL PROPERTIES OF Tl-O MONOLAYER SUPERCONDUCTORS: $\text{TlBa}_{1+x}\text{La}_{1-x}\text{CuO}_5$ AND $\text{TlBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$

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**ABSTRACT** Two systems of Tl-based copper oxides with a Tl-O monolayer structure,  $\text{TlBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$  (1212 phase) and  $\text{TlBa}_{1+x}\text{La}_{1-x}\text{CuO}_5$  (1201 phase), have been synthesized. For  $x=0$ , where formal Cu valences in both systems were estimated to be just +2, samples exhibited semiconductive or insulating behaviors in the resistivity measurement. With increasing  $x$ , electrical properties of samples became more metallic and superconductivity appeared. For the 1201 phase, the maximum  $T_c$  value was about 40K at around  $x=0.2$ , and sample became a metallic nonsuperconductor for  $x>0.4$ . For the 1212 phase, maximum  $T_c$  value was about 90K at around  $x=0.8$  and metallic nonsuperconductivity was not observed.

### 1. INTRODUCTION

Tl-based superconductors with a general composition of  $\text{Tl}_m\text{Ba}_2\text{Ca}_{n-1}\text{Cu}_n\text{O}_{2+2n+m}$  were specified by the number of Tl-O sheets ( $m=1,2$ ) and the number of  $\text{CuO}_2$  planes ( $n=1,2,3,4,\dots$ ) in a formula unit. In the earliest stage, single phase samples which contained one to three  $\text{CuO}_2$  planes between double Tl-O layers ( $m=2$ ) were synthesized, and both electrical properties and crystal structures were closely examined[1]. Whereas, it was rather difficult to obtain a single phase sample with Tl-O monolayer sheets ( $m=1$ ), especially for  $n=1$  or 2. The difficulty in getting single phase samples of Tl-O monolayer compounds seems to be related to the high formal oxidation state of Cu. For example, the formal Cu valence in  $\text{TlBa}_2\text{CuO}_5$  is as high as +3 which is much higher than that in the ordinary copper oxides.

In our previous study, we could have prepared easily a single phase sample of  $\text{TlBa}_2\text{YCu}_2\text{O}_7$  by substituting trivalent Y for divalent Ca in  $\text{TlBa}_2\text{CaCu}_2\text{O}_7$  (1212 phase)[2]. This means that the 1212 structure is stabilized by such substitution that decreases the formal copper valence. In the present study, we synthesized two substituted systems having Tl-O monolayer

sheets,  $\text{TlBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$  (1212) and  $\text{TlBa}_{1+x}\text{La}_{1-x}\text{CuO}_5$  (1201), in which the formal Cu valence was reduced to +2 for  $x=0$ . Electrical and superconducting properties and the phase stability will be discussed with respect to the composition  $x$ .

## 2. EXPERIMENTAL

Samples were prepared by a solid state reaction as described in [2,3]. DC resistivities and AC magnetic susceptibilities were measured using a conventional four probe method and a self inductance method, respectively. Crystal structures and chemical compositions were studied by the powder X-ray diffraction (XRD) and the electron-probe micro-analysis (EPMA).

## 3. RESULTS AND DISCUSSION

### 3.1 $\text{TlBa}_{1+x}\text{La}_{1-x}\text{CuO}_5$ system

According to XRD, no extra peaks due to impurity phases were found for  $0 < x < 0.4$ , while  $(\text{La},\text{Ba})_2\text{CuO}_4$  phase and a trace of  $\text{Tl}_2\text{Ba}_2\text{CuO}_6$  phase were observed for  $x < 0$  and for  $x > 0.4$ , respectively. On the other hand, EPMA measurement revealed that the Ba contents were decreased to less than the nominal values for  $x > 0$  as shown in Table 1. This suggests the appearance of a small amount of second phases except for samples with  $x=0$ . However, the quantities of the second phases estimated from XRD measurement were no more than a few percents, so the samples could be regarded as close to single phase. In the following discussion, the real  $x$  value determined by EPMA will be used. Samples with negative  $x$  values could not have been obtained in the present experiment, which suggested that the solubility limit of La for  $\text{TlBa}_2\text{CuO}_5$  was located at around  $x=0$  where the formal Cu valence was just +2. The lattice parameters  $a_0$  and  $c_0$  are plotted against  $x$  in Fig.1(a). A positive correlation was observed between  $c_0$  and  $x$ , whereas  $a_0$  exhibited no obvious dependence on  $x$ . Similar tendencies were observed in  $\text{TlBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$  [Fig.1(b)] and  $\text{Bi}_2\text{Sr}_2\text{Ca}_{1-x}\text{Y}_x\text{Cu}_2\text{O}_8$  [4].

$\text{TlBaLaCuO}_5$ ,  $x=0$  sample, exhibits a semiconductor-like behavior in resistivity. Since the formal Cu valence of this compound is +2, the hole concentration was estimated to be 0. It was expected that the Ba substitution

Table 1 Chemical compositions of samples determined by EPMA with mixing compositions of  $\text{TlBa}_{1+x}\text{La}_{1-x}\text{CuO}_5$ .

x	Tl	Ba	La	Cu
-0.2	0.93	1.00	1.00	1.02
0.0	0.91	1.01	0.99	1.09
0.1	0.94	1.06	0.94	1.06
0.2	0.98	1.11	0.89	1.05
0.3	0.96	1.12	0.88	1.04
0.4	0.99	1.19	0.81	1.06
0.5	0.98	1.36	0.64	1.04
0.6	1.00	1.40	0.60	1.04

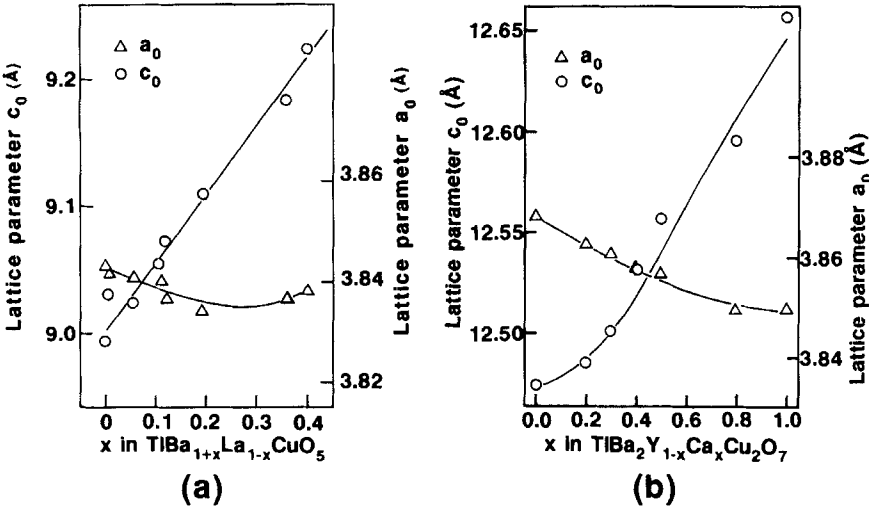


Fig. 1 The tetragonal lattice parameters  $a_0$  and  $c_0$  are plotted against  $x$  for (a)  $\text{TlBa}_{1+x}\text{La}_{1-x}\text{CuO}_5$  and (b)  $\text{TlBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$

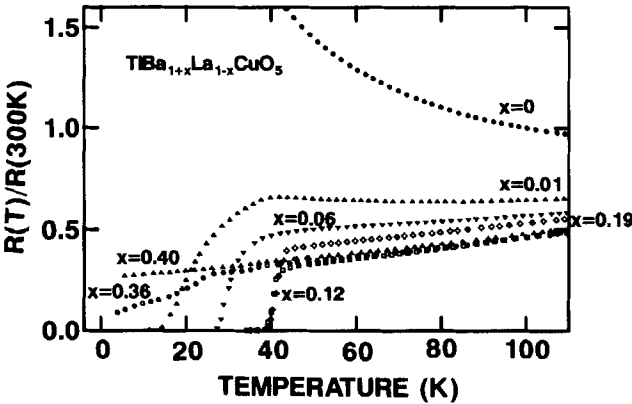


Fig. 2 Normalized resistivities plotted against temperature for  $\text{TlBa}_{1+x}\text{La}_{1-x}\text{CuO}_5$ .

for La would dope hole carriers in the system. The temperature dependencies of normalized resistivities were shown in Fig.2. With increasing  $x$ , the temperature dependence of the resistivity became more metallic. For  $0 < x < 0.4$ , superconductivity appeared and the  $T_c$  showed a maximum value of 40K at around  $x = 0.2$ . For  $x \geq 0.4$ , the sample became a metallic nonsuperconductor. Such correlation between hole concentration and  $T_c$  was very similar to that observed in La-Sr-Cu-O system[5].

It should be noted that not only  $T_c$  but also the superconducting volume fraction changed with  $x$ . AC susceptibilities at 4.2K, which indicated the superconducting volume fraction, and the  $T_c$ 's determined by resistivity measurement are plotted against  $x$  in Fig.3. It was clear that the change in  $T_c$  was accompanied by the change in the superconducting volume fraction. It was suggested that good superconductivity ( $T_c = 40$ K) occurred only in a very narrow range of composition around  $x = 0.2$ .

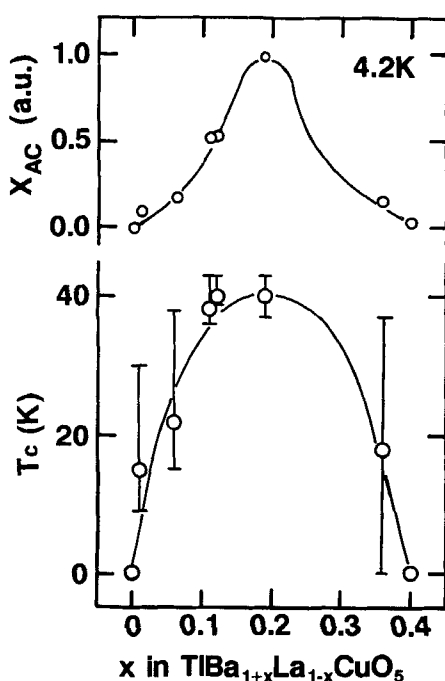


Fig. 3 AC susceptibilities at 4.2K and  $T_c$  determined by DC resistivities are plotted against  $x$  value. Upper bar, circle, lower bar represent onset, midpoint, zero resistivity temperature, respectively.

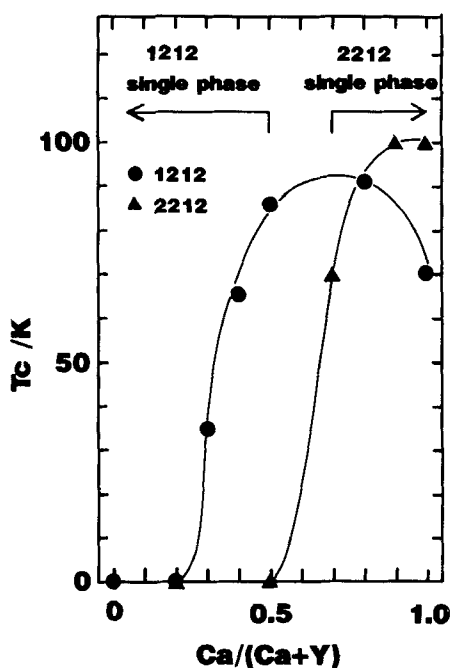


Fig. 4  $T_c$ 's and the single phase regions for samples with starting compositions of  $\text{TlBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$  (1212, circles) and  $\text{Tl}_2\text{Ba}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_8$  (2212, triangles).

### 3.2 $\text{TlBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$ system

When the starting composition was 1212, single phase samples with 1212 structure were obtained for  $0 < x < 0.5$ , while  $\text{BaCuO}_2$  and a trace of 2212 phases appeared for  $x \geq 0.5$ . This phase instability for larger  $x$  values seems to be due to the highly oxidized Cu state, +2.5 for  $x=1$ . In contrast, when the starting composition was 2212, we obtained single phase 2212 samples only for  $0.7 < x < 1.0$ , and 1212 phase began to appear for smaller  $x$  values. This suggested that the 2212 structure became unstable when the formal Cu valence was too much decreased from +2. Therefore, for all systems of the present study with 1201, 1212 and 2212 compositions, single phase samples were obtained only when the formal Cu valence was near +2.

Fig.4 shows  $T_c$ 's and single phase regions for both cases where the samples were synthesized with starting compositions of 1212 and 2212, respectively. Although the samples were not single phase in the large  $x$  region for 1212 composition and the small  $x$  region for 2212 composition, the main impurity phase was  $\text{BaCuO}_2$  which was an insulator. So the superconducting properties in Fig. 4 are considered to be very close to real behavior. For 1212 system, the  $x=0$  sample ( $\text{TlBa}_2\text{YCu}_2\text{O}_7$ ) was an insulator with an antiferromagnetic order[6]. As  $x$  increased, hole carriers were introduced and superconductivity appeared for  $x > 0.3$ .  $T_c$  showed a maximum value of about 90K but metallic nonsuperconducting state was not achieved even for  $x=1$ . For 2212 system, the  $x=1$  sample exhibited 100K superconductivity and the  $T_c$  decreased with  $x$  decreased.

In conclusion, We have prepared two series of Tl-based copper oxides with a Tl-O monolayer structure;  $\text{TlBa}_{1+x}\text{La}_{1-x}\text{CuO}_5$  (1201 phase) and  $\text{TlBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$  (1212 phase). For 1201 phase superconductivity appeared for  $0 < x < 0.4$ , and maximum  $T_c$  value was about 40K at around  $x=0.2$ . For 1212 phase, superconductivity appeared for  $x > 0.3$ , and maximum  $T_c$  value was about 40K at around  $x=0.8$ . It is concluded that superconductivity appears at compositions where the hole concentration becomes an optimum value.

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